

LET THERE BE LIGHT

Een verhandeling over straling

François van Langevelde

The Creation of Light (Genesis 1.3-5)



Gustave Doré – Bible Illustrations

1 In den beginne schiep God den hemel en de aarde.

2 De aarde nu was woest en ledig, en duisternis was op den afgrond; en de Geest Gods zweefde op de wateren.

3 En God zeide: Daar zij licht! en daar werd licht.

4 En God zag het licht, dat het goed was; en God maakte scheiding tussen het licht en tussen de duisternis.

5 En God noemde het licht dag, en de duisternis noemde Hij nacht. Toen was het avond geweest, en het was morgen geweest, de eerste dag.

Aldus De Statenvertaling 1637

The Four Fundamental Forces



Gravity

Gravitonen (?)

1



**Electromagnetic
Force**

Fotonen

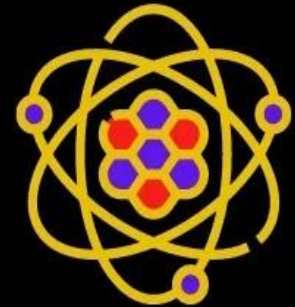
10^{38}



Weak Force

W&Z Bosonen

10^{15}

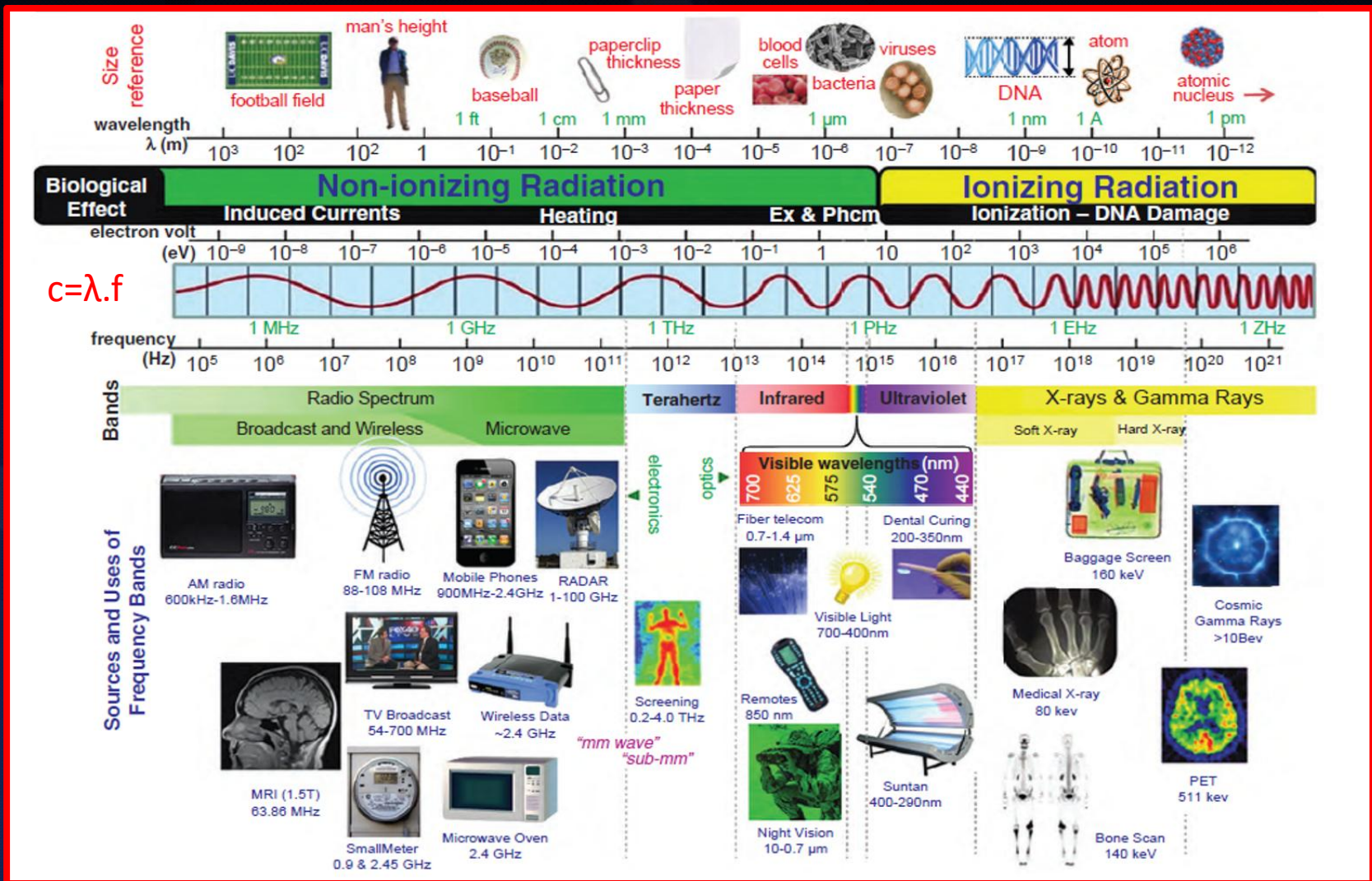


Strong Force

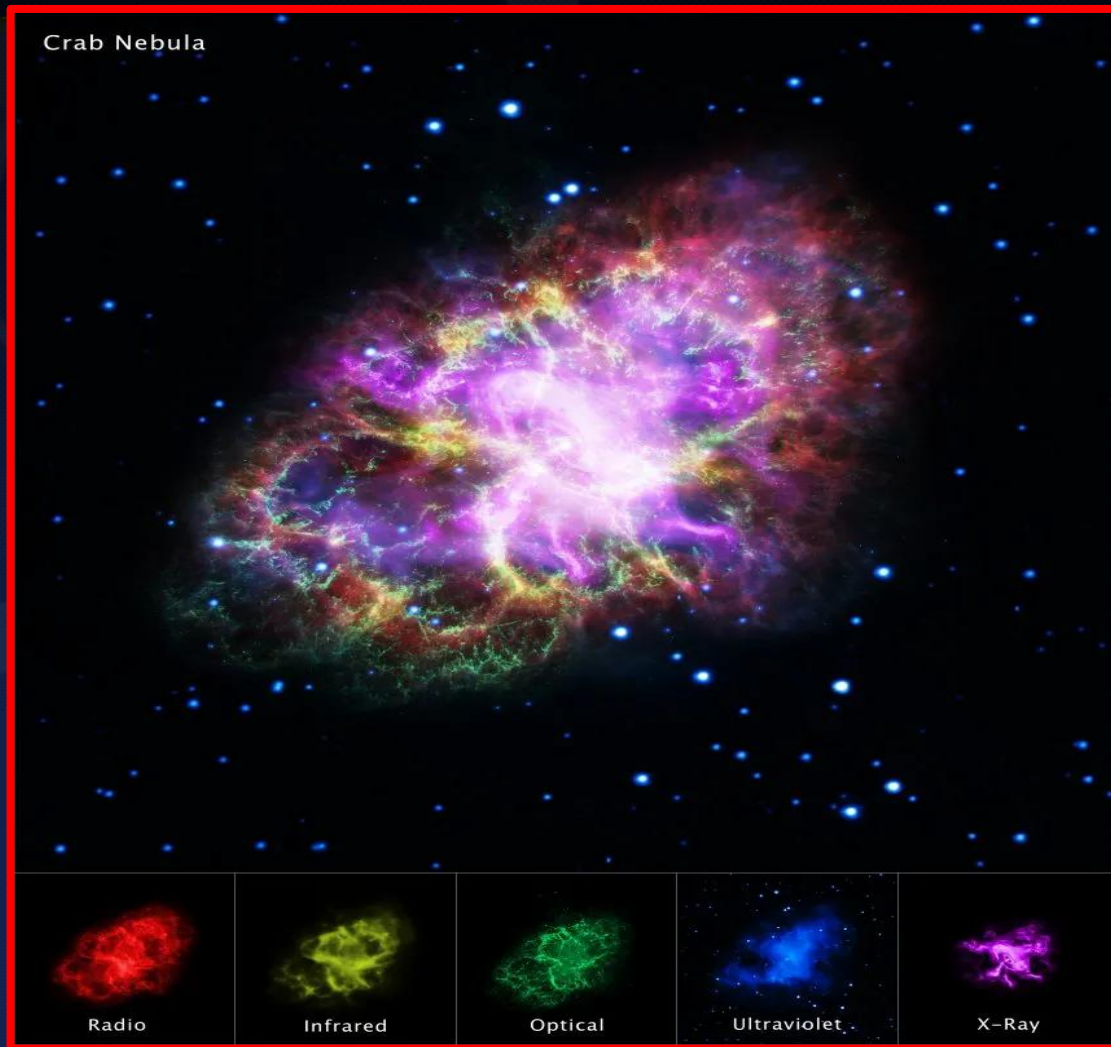
Gluonen

10^{40}

The Electromagnetic Spectrum I



The Electromagnetic Spectrum II

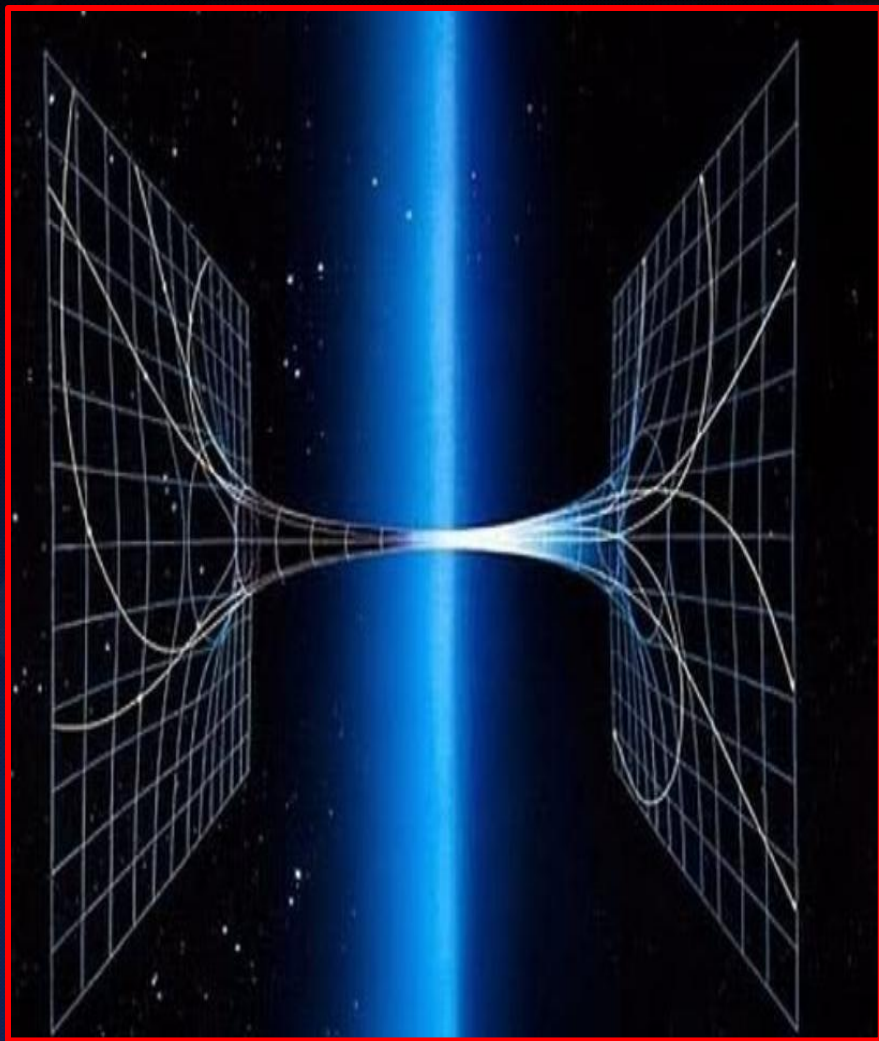


LET THERE BE LIGHT

Een verhaal over straling

François van Langevelde

"Genesis" according to George Gamow



From Big Crunch to Big Bang?

In het begin schiep God Straling en Ylem.
En Ylem was zonder vorm of getal, en de nucleonen
raasden als gekken over het oppervlak van het diepe.

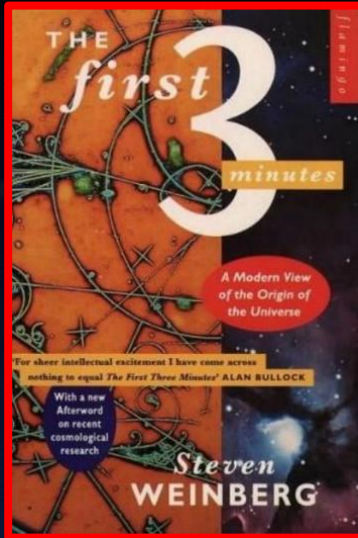
En God zei: "Laat er massa twee zijn."
En er was massa twee.
En God zag Deuterium, en het was goed.

En God zei: "Laat er massa drie zijn."
En er was massa drie.
En God zag Tritium, en het was goed.

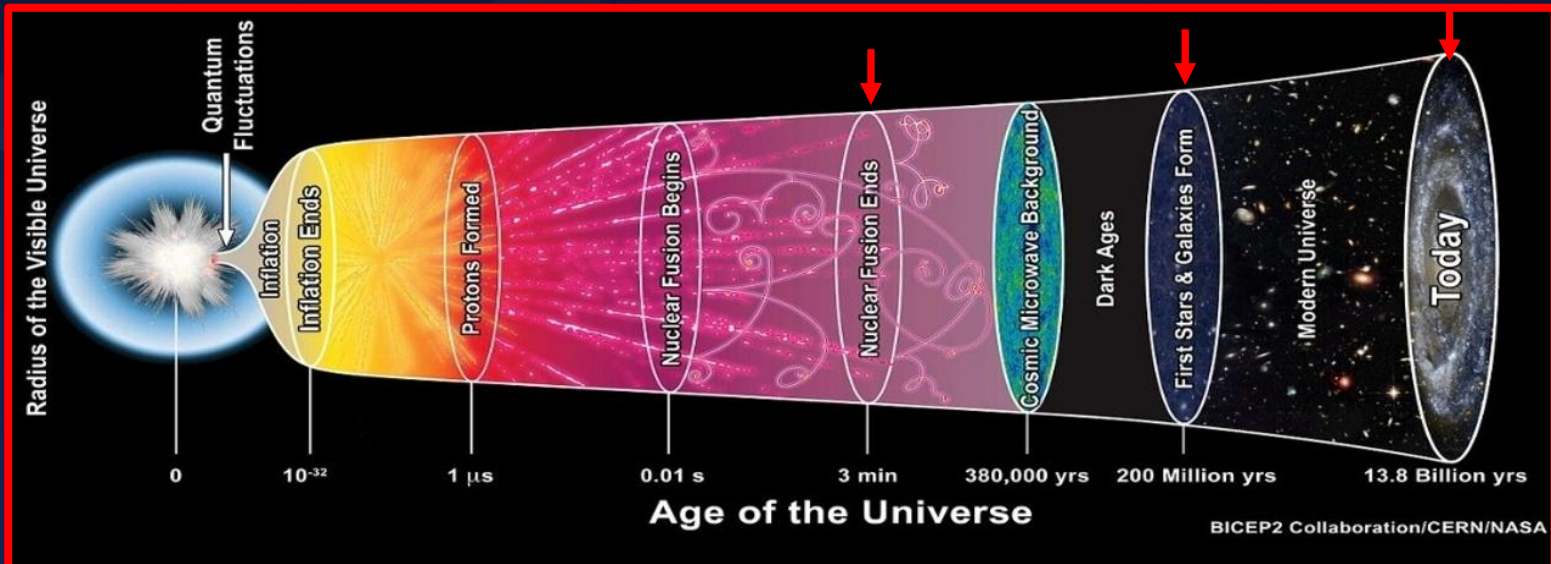
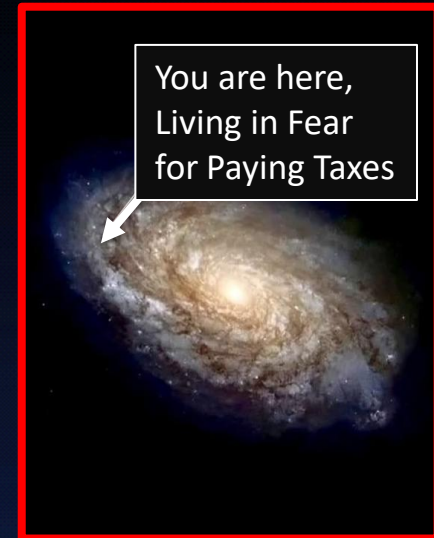
En God bleef getallen noemen totdat Hij bij de
Transuranium elementen kwam.
Maar toen Hij terugkeek op zijn werk, zag
Hij dat het niet goed was.
In de opwindning van het tellen had
Hij gemist om massa vijf te noemen, en dus
konden er natuurlijk geen zwaardere elementen
gevormd zijn.

.....

First Three Minutes and beyond . . .



										F-14	F-15	F-16	F-17	F-18	F-19
										5.0E-22s	4.1E-22s	1.1E-19s	1.075m	1.830h	100
										O-12	O-13	O-14	O-15	O-16	O-17
										5.63E-21s	8.58ms	1.1770m	2.037m	99.757	0.038
										N-10	N-11	N-12	N-13	N-14	N-15
										2.0E-22s	5.5E-22s	11.000ms	9.965m	0.364	7.13s
										C-8	C-9	C-10	C-11	C-12	C-13
										2.0E-21s	126.5ms	19.305s	20.364m	98.95	1.07
										B-7	B-8	B-9	B-10	B-11	B-12
										3.3E-22s	770ms	8.5E-19s	19.9	80.1	20.20ms
										Be-6	Be-7	Be-8	Be-9	Be-10	Be-11
										5.0E-21s	53.22d	8.2E-17s	100	1.51E6y	13.76ms
										Li-4	Li-5	Li-6	Li-7	Li-8	Li-9
										7.6E-23s	3.7E-22s	7.59	92.41	838.75ms	178.3ms
										He-3	He-4	He-5	He-6	He-7	He-8
										0.000134	99.999866	7.0E-22s	806.82ms	3.0E-21s	119.1ms
										H-1	H-2	H-3	H-4	H-5	H-6
										99.985	0.0115	12.32y	1.4E-22s	2.4E-22s	2.9E-22s
										n-1	no stable nuclide 5				
										10.25m					



Human Elemental Composition

Human Body Ingredients

The four ingredients below are essential parts of the body's protein, carbohydrate and fat architecture.

O

OXYGEN

65.0%

Critical to the conversion of food into energy.

C

CARBON

18.5%

The so-called backbone of the building blocks of the body and a key part of other important compounds, such as testosterone and estrogen.

H

HYDROGEN

9.5%

Helps transport nutrients, remove wastes and regulate body temperature. Also plays an important role in energy production.

N

NITROGEN

3.3%

Found in amino acids, the building blocks of proteins; an essential part of the nucleic acids that constitute DNA.

(Percentage of body weight. Source: *Biology*, Campbell and Reece, eighth edition.)

Other Key Elements

Calcium 1.5%
Lends rigidity and strength to bones and teeth; also important for the functioning of nerves and muscles, and for blood clotting.

Phosphorus 1.0%
Needed for building and maintaining bones and teeth; also found in the molecule ATP (adenosine triphosphate), which provides energy that drives chemical reactions in cells.

Potassium 0.4%
Important for electrical signaling in nerves and maintaining the balance of water in the body.

Sulfur 0.3%
Found in cartilage, insulin (the hormone that enables the body to use sugar), breast milk, proteins that play a role in the immune system, and keratin, a substance in skin, hair and nails.

Chlorine 0.2%
Needed by nerves to function properly; also helps produce gastric juices.

Sodium 0.2%
Plays a critical role in nerves' electrical signaling; also helps regulate the amount of water in the body.

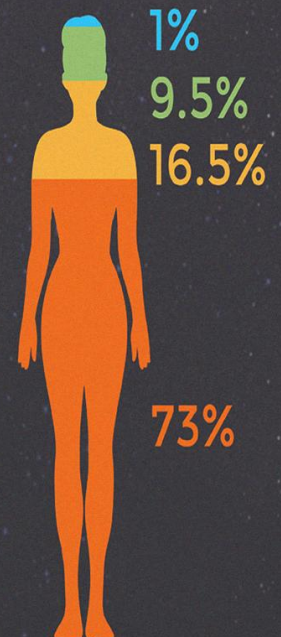
Magnesium 0.1%
Plays an important role in the structure of the skeleton and muscles; also found in molecules that help enzymes use ATP to supply energy for chemical reactions in cells.

Iodine (trace amount)
Part of an essential hormone produced by the thyroid gland; regulates metabolism.

Iron (trace amount)
Part of hemoglobin, which carries oxygen in red blood cells.

Zinc (trace amount)
Forms part of some enzymes involved in digestion.

Elements from Big Bang and Stars



Radionuclides in the body:

^{40}K isotope of Potassium

^{14}C isotope of Carbon

^{226}Ra isotope of Radium

Earliest Human Experience with Light

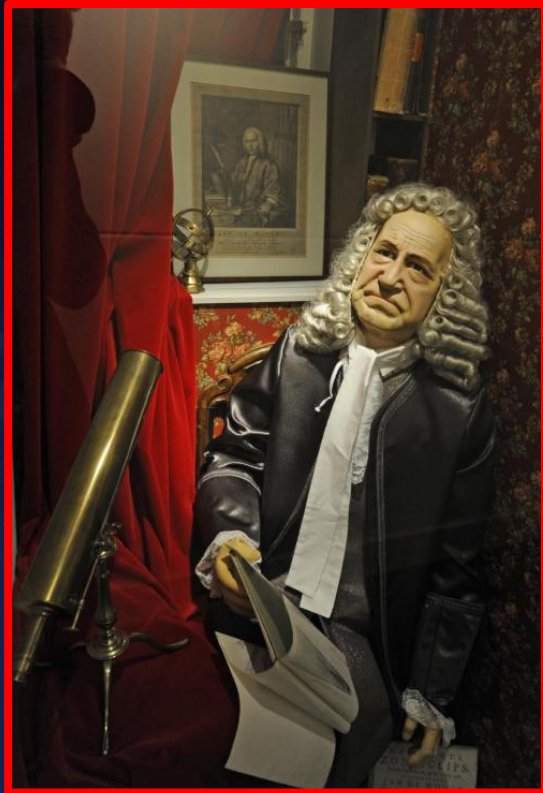
SUN



FIRE



17th Century Age of Light (Telescope)

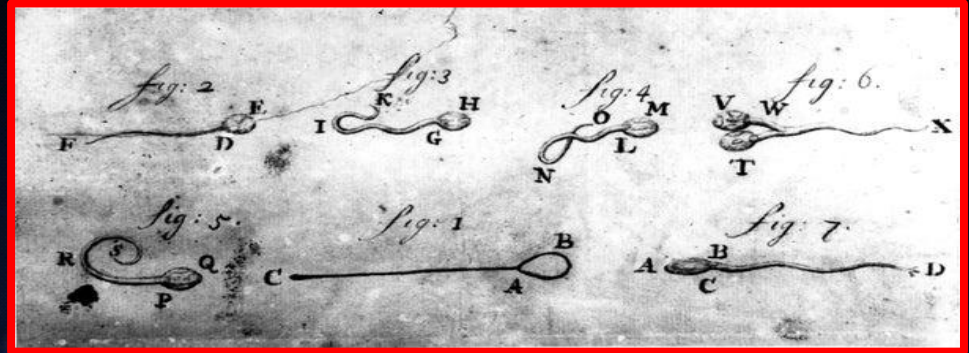
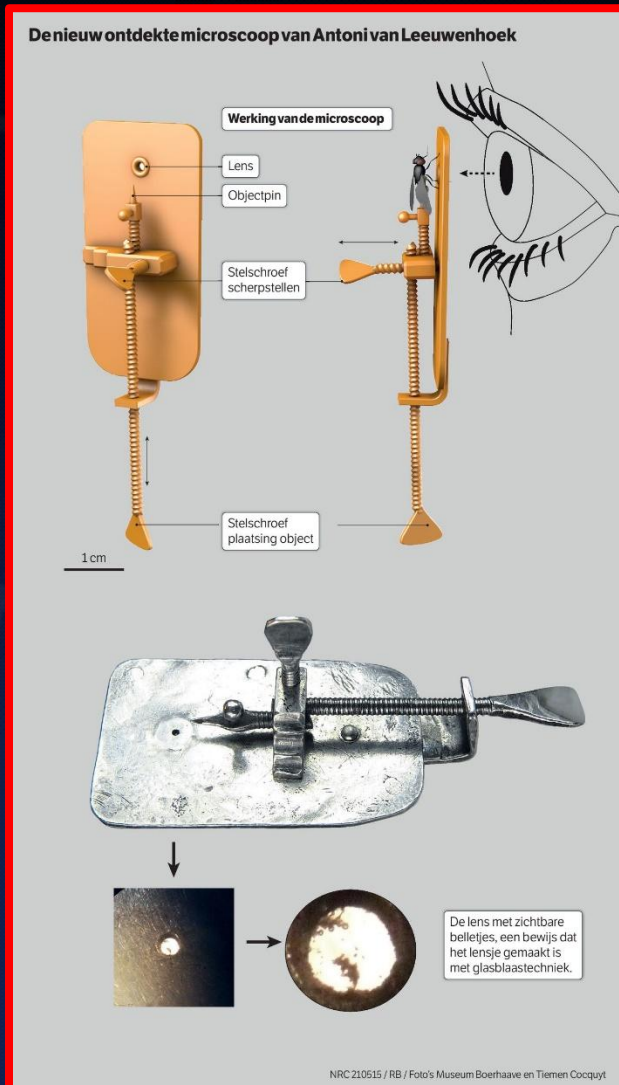


Johannes Lipperhey



1608

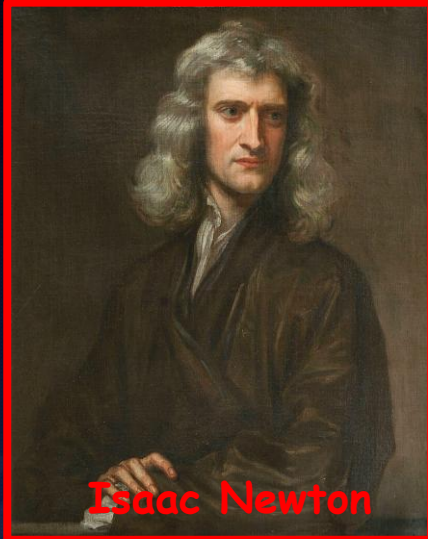
17th Century Age of Light (Microscope)



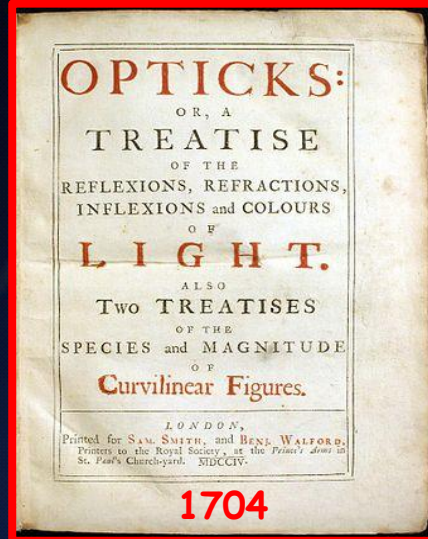
Antoni van Leeuwenhoek 1632-1723



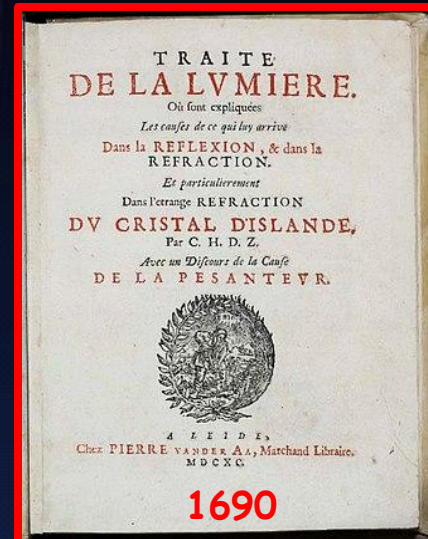
17th Century Age of Light (particles or waves?)



Isaac Newton



1704



1690



Christiaan Huygens

Newton

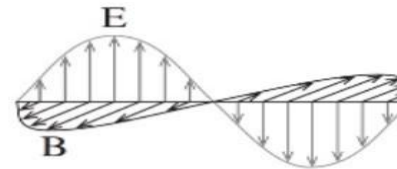
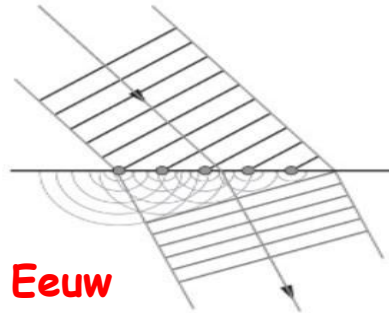
Huygens

Maxwell

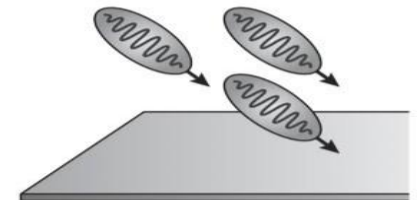
Einstein



17e Eeuw



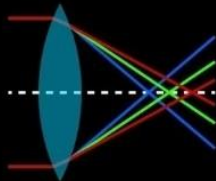
19e Eeuw



20e Eeuw

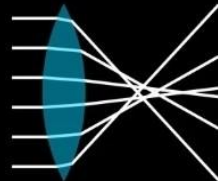
18th Century Age of Light Optics

CHROMATIC ABERRATION

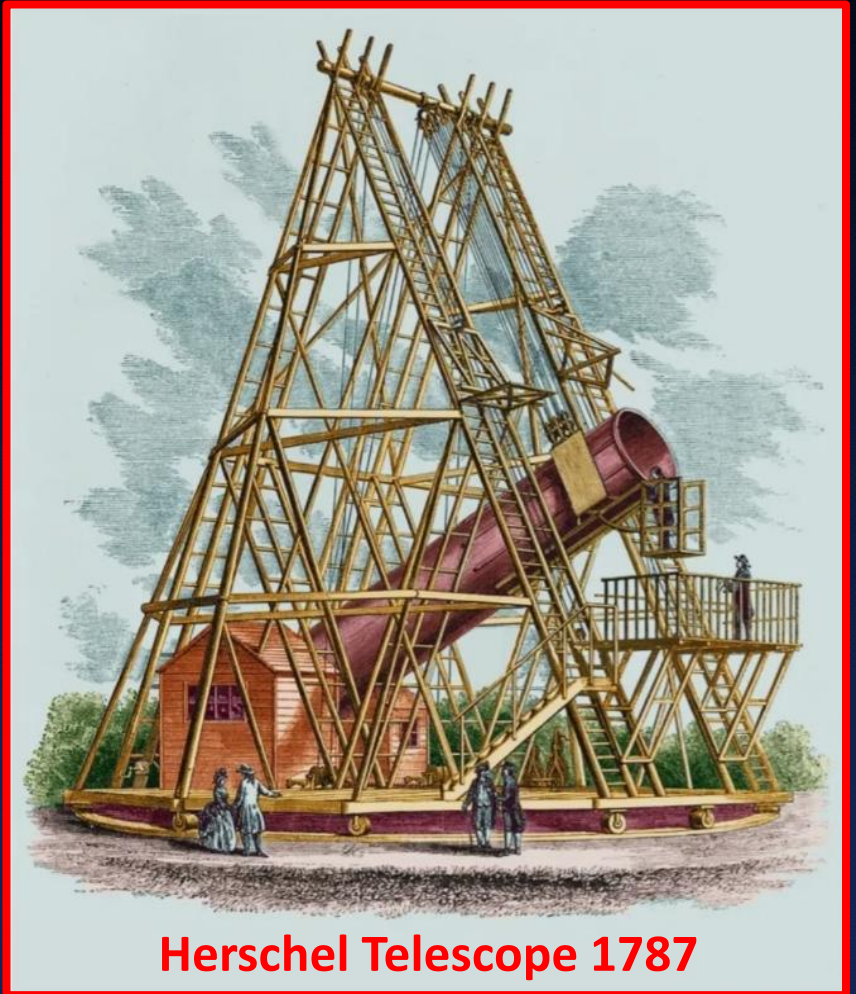


occurs because the lens refracts the various colors at different angle according to the wavelength

SPHERICAL ABERRATION

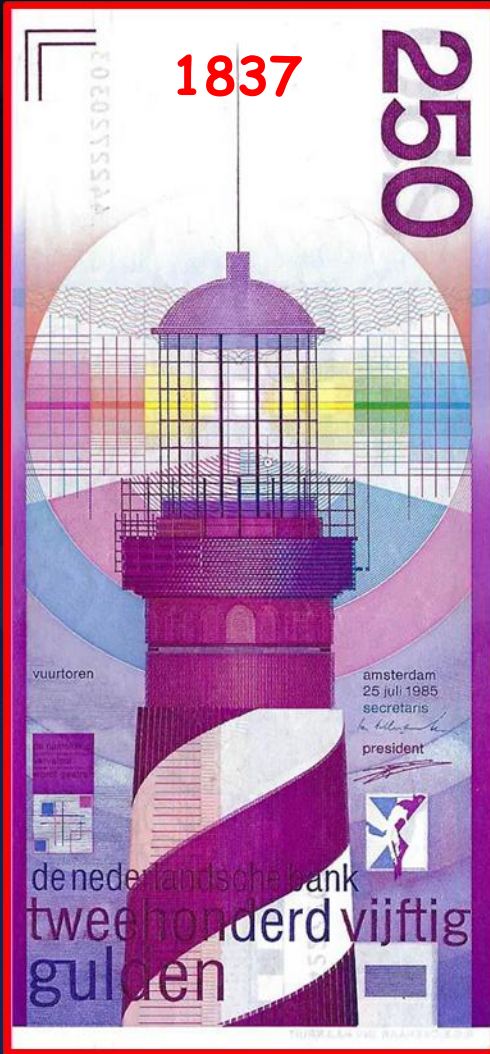


occurs when light waves passing through a lens with spherical surface are not brought into exact focus



Herschel Telescope 1787

19th Century Age of Light/Electricity



Davy
Ørsted
Volta
Ampere
Faraday
Ohm
Henry
Morse
Maxwell
Bell
Edison
Thompson

Davy Lamp



1815

Booglamp
Electromotor
Electromagneet
Dynamo
Telegraafkabel
Brandstofcel
Oplaadbare batterij
EM theorie
Gloeilamp
Transformator
Radiogolven
Draadloze telegrafie
Kathodestraalbuis
Ontdekking electron

Crookes Tube



1879

New York City 1880 vs Bangkok 2025



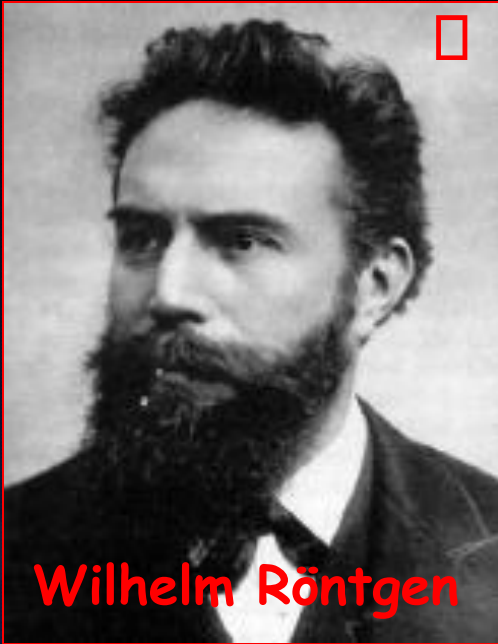
Dit is New York City in de jaren 1880.



Dit is Bangkok anno 2025.

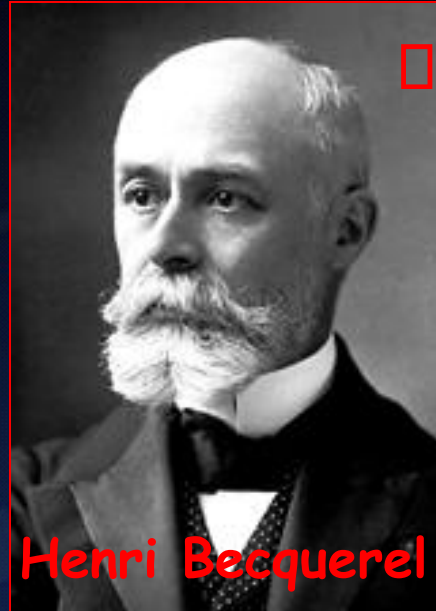
Discoveries of X-rays & Radioactivity

□ 1895 X-rays



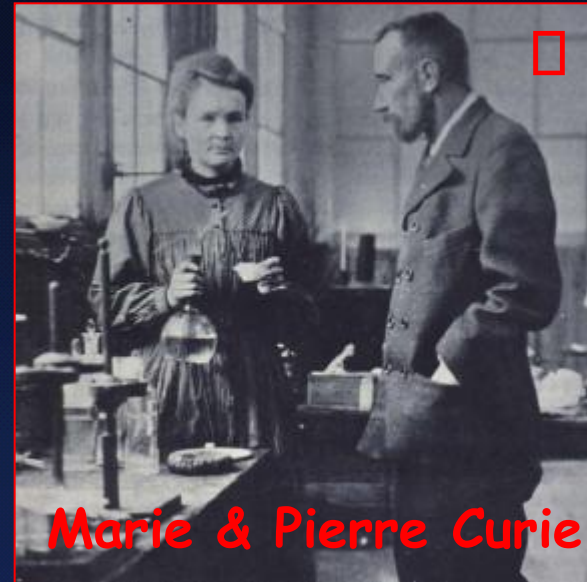
Wilhelm Röntgen

□ 1896 Radioactivity

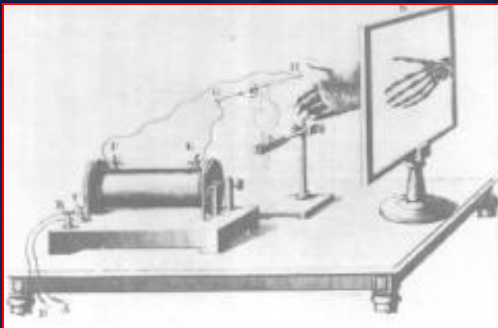


Henri Becquerel

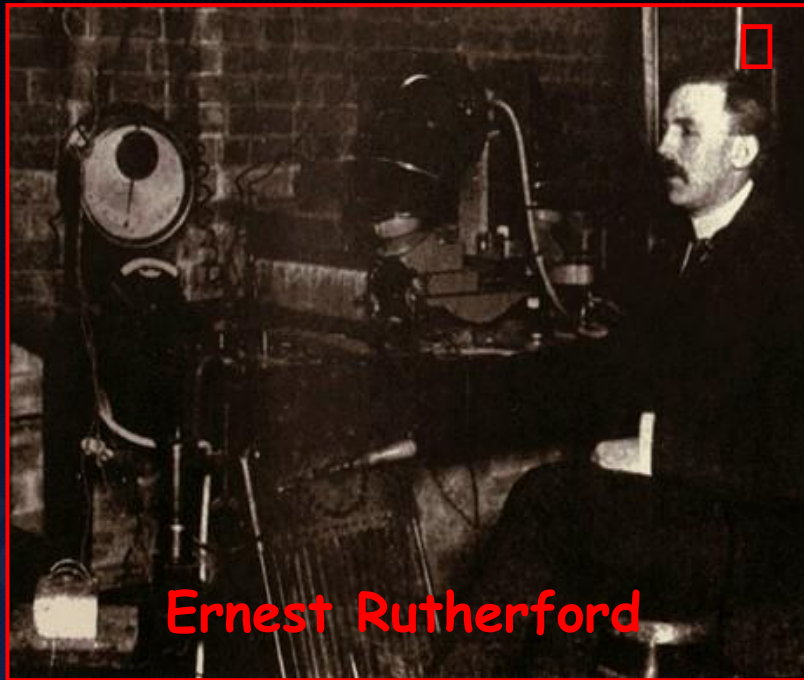
□ 1898
Radio-
active
Elements
(Po & Ra)



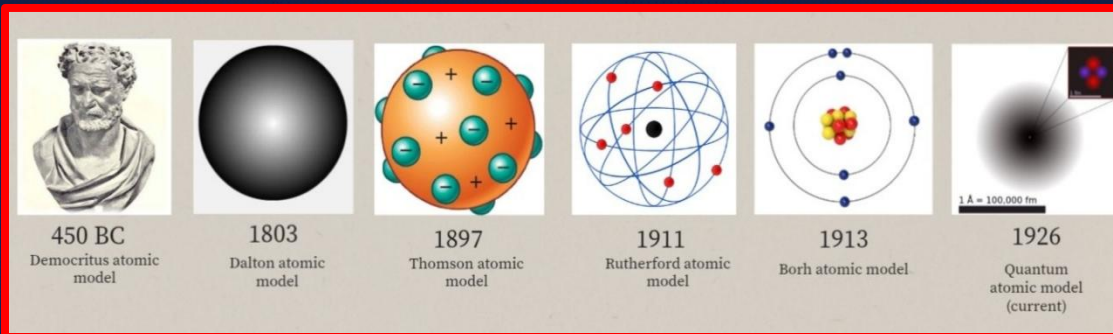
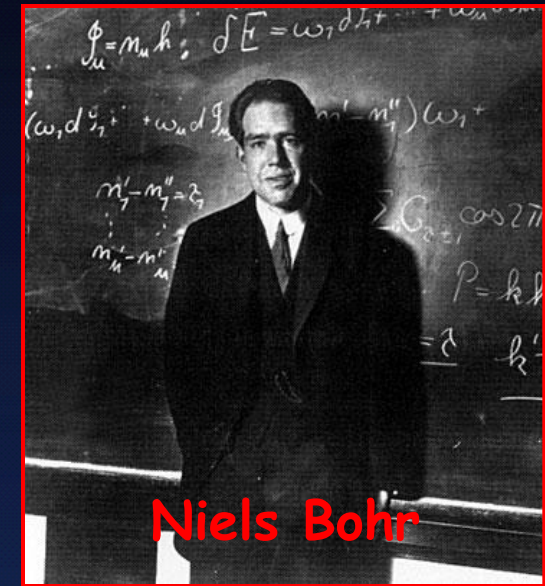
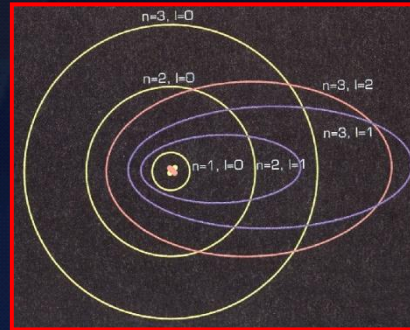
Marie & Pierre Curie



Birth of Atom Physics



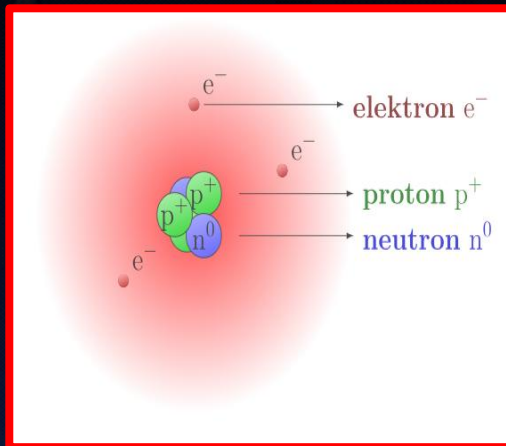
1919 nuclear reaction by using radio-active elements



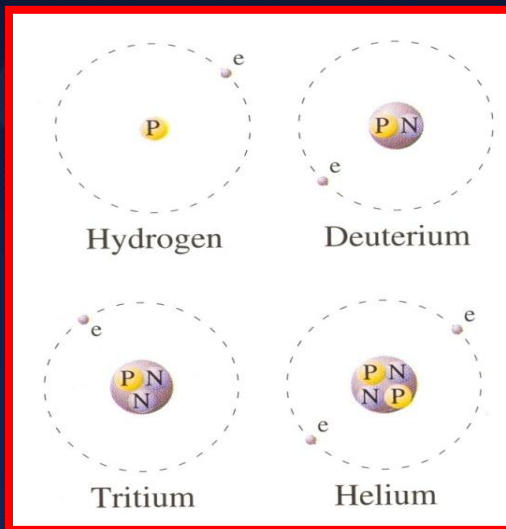
1913 Atom Models



The Composition of Matter



- the Elements
118 elements
atoms and nuclei
periodical system
(Mendeleev)



- the Nuclides
280 stable
>2300 instable
nuclide chart
(Segrè)

Alkali metals	Halogens
Alkaline-earth metals	Noble gases
Transition metals	Rare-earth elements (21, 39, 57–71) and lanthanoid elements (57–71 only)
Other metals	
Other nonmetals	Actinoid elements

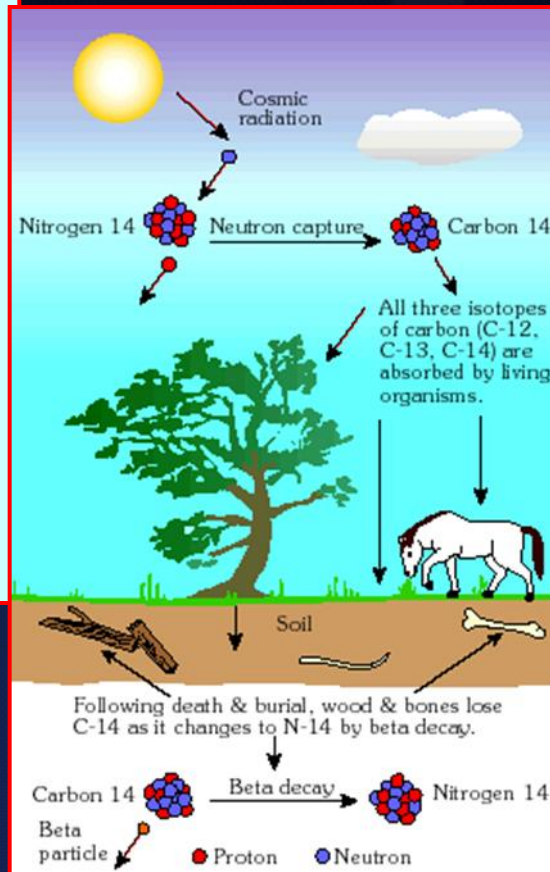
lanthanoid series. 6actinoid series 7

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Natural and Other Radionuclides

URANIUM 238 (U238) RADIOACTIVE DECAY

type of radiation	nuclide	half-life
	uranium—238	4.5×10^9 years
α	thorium—234	24.5 days
β	protactinium—234	1.14 minutes
β	uranium—234	2.33×10^5 years
α	thorium—230	8.3×10^4 years
α	radium—226	1590 years
α	radon—222	3.825 days
α	polonium—218	3.05 minutes
α	lead—214	26.8 minutes
β	bismuth—214	19.7 minutes
β	polonium—214	1.5×10^{-4} seconds
α	lead—210	22 years
β	bismuth—210	5 days
β	polonium—210	140 days
α	lead—206	stable



□ Terrestrial radionuclides:
 $^{238}\text{U}(4.5 \times 10^9 \text{y})$, $^{232}\text{Th}(1.4 \times 10^{10} \text{y})$,
 $^{40}\text{K}(1.3 \times 10^9 \text{y})$, $^{235}\text{U}(7 \times 10^8 \text{y})$, ...

□ Cosmogene radionuclides:
 $^3\text{H}(12,4 \text{y})$, $^7\text{Be}(53 \text{d})$, $^{14}\text{C}(5730 \text{y})$,
 $^{22}\text{Na}(2,6 \text{y})$, ...

□ Artificial Produced
 Radionuclides
 medical:

$^{18}\text{F}(2 \text{h})$, $^{99\text{m}}\text{Tc}(6 \text{h})$, $^{131}\text{I}(8 \text{d})$, ...

industrial:

$^{60}\text{Co}(5,7 \text{y})$, $^{241}\text{Am}(432 \text{y})$, ...

Human Elemental Composition

Human Body Ingredients

The four ingredients below are essential parts of the body's protein, carbohydrate and fat architecture.

O

OXYGEN

65.0%

Critical to the conversion of food into energy.

C

CARBON

18.5%

The so-called backbone of the building blocks of the body and a key part of other important compounds, such as testosterone and estrogen.

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NITROGEN

3.3%

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(Percentage of body weight. Source: *Biology*, Campbell and Reece, eighth edition.)

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Important for electrical signaling in nerves and maintaining the balance of water in the body.

Sulfur 0.3%
Found in cartilage, insulin (the hormone that enables the body to use sugar), breast milk, proteins that play a role in the immune system, and keratin, a substance in skin, hair and nails.

Chlorine 0.2%
Needed by nerves to function properly; also helps produce gastric juices.

Sodium 0.2%
Plays a critical role in nerves' electrical signaling; also helps regulate the amount of water in the body.

Magnesium 0.1%
Plays an important role in the structure of the skeleton and muscles; also found in molecules that help enzymes use ATP to supply energy for chemical reactions in cells.

Iodine (trace amount)
Part of an essential hormone produced by the thyroid gland; regulates metabolism.

Iron (trace amount)
Part of hemoglobin, which carries oxygen in red blood cells.

Zinc (trace amount)
Forms part of some enzymes involved in digestion.

Elements from Big Bang and Stars



1%

9.5%

16.5%

73%

Radionuclides in the body:

^{40}K isotope of Potassium

^{14}C isotope of Carbon

^{226}Ra isotope of Radium

A

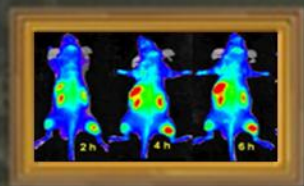
PICTURE



IS
WORTH



WORDS



1000





CT scanner



Ultrasound scanner



PET scanner



SPECT Camera



MRI scanner

Medical Imaging Modalities

- Non-ionizing Radiation Techniques

MR Techniques (MRI)

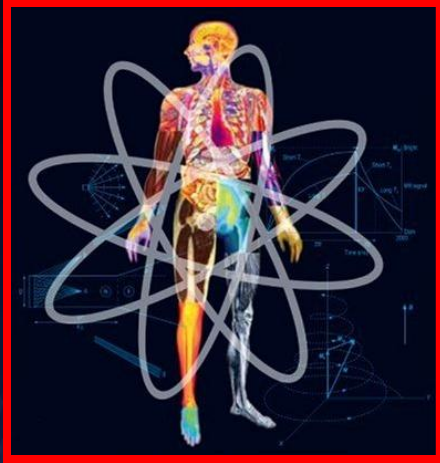
Ultrasound

- Ionizing Radiation Techniques

X-ray Techniques (CT)

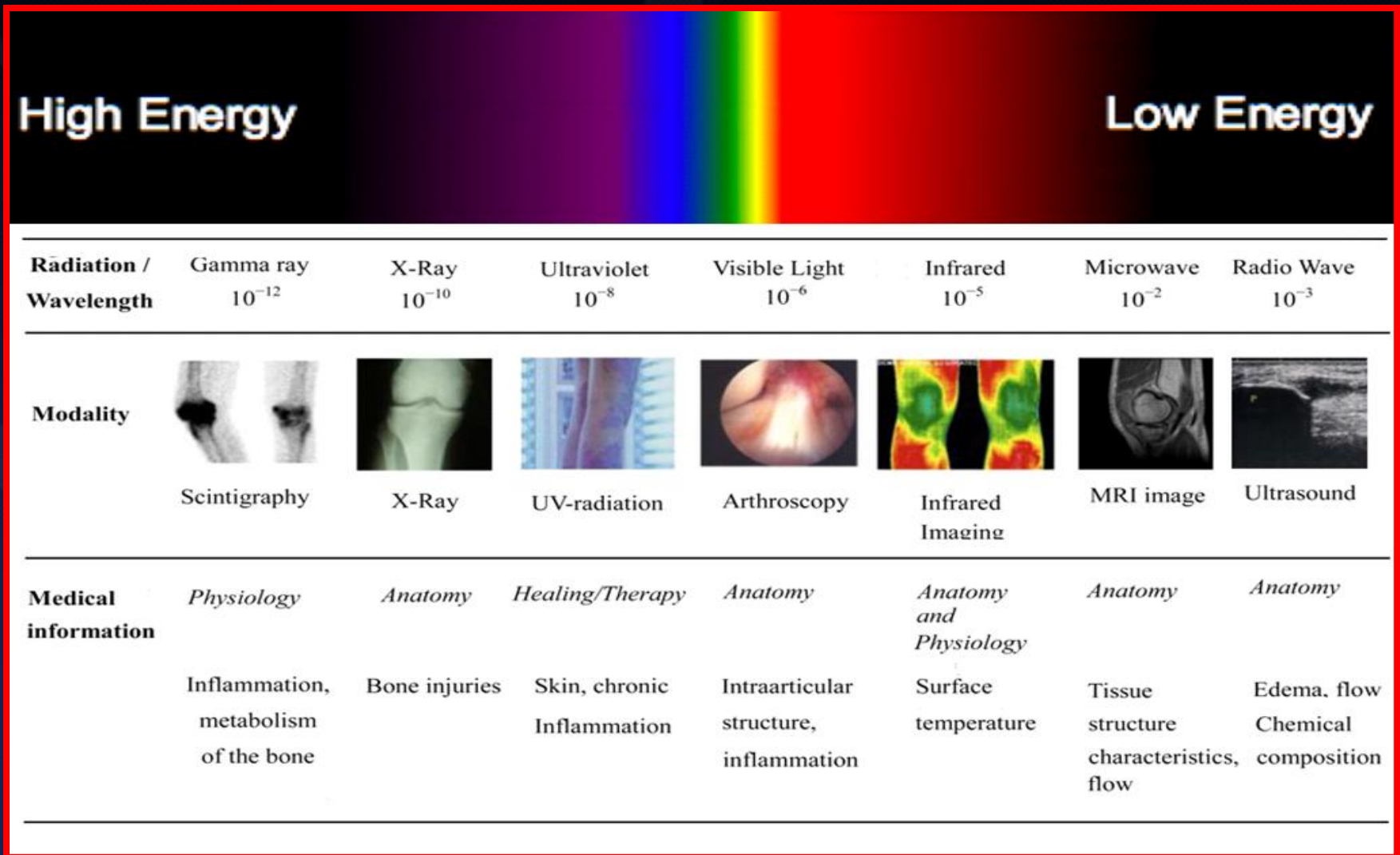
Nuclear Imaging (PET)

Why are there so many modalities?

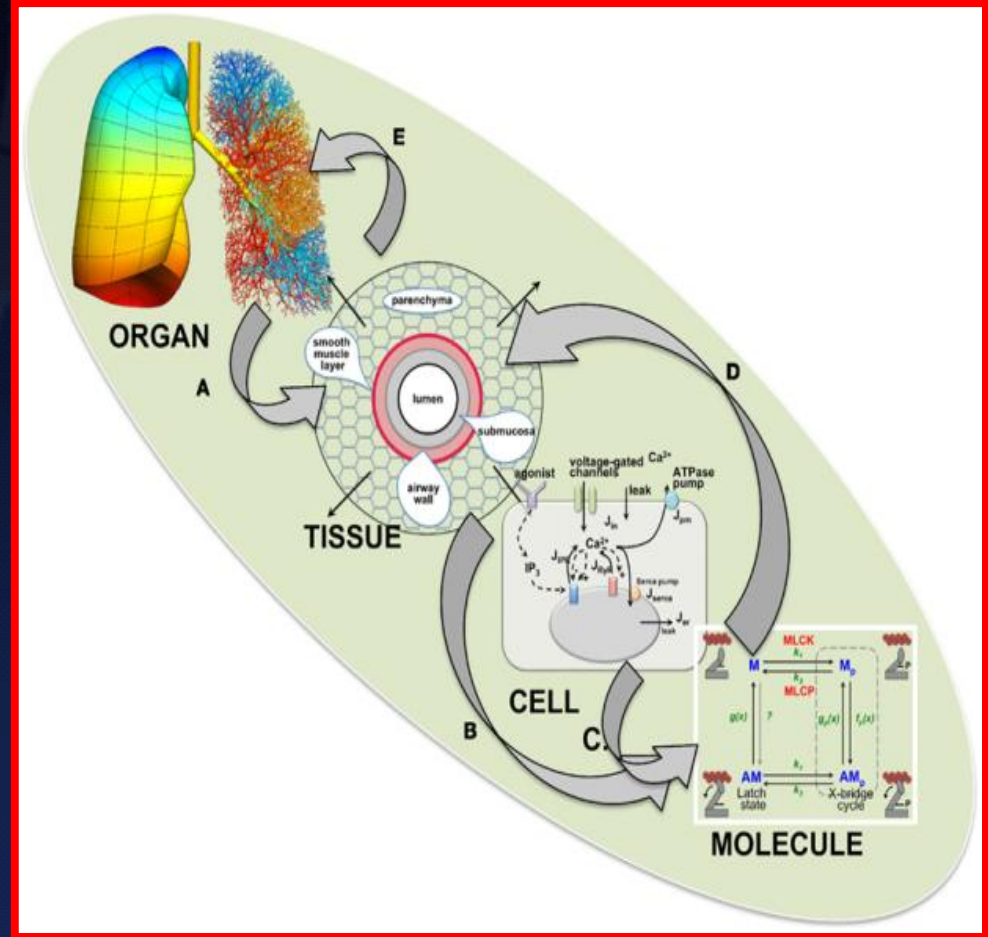
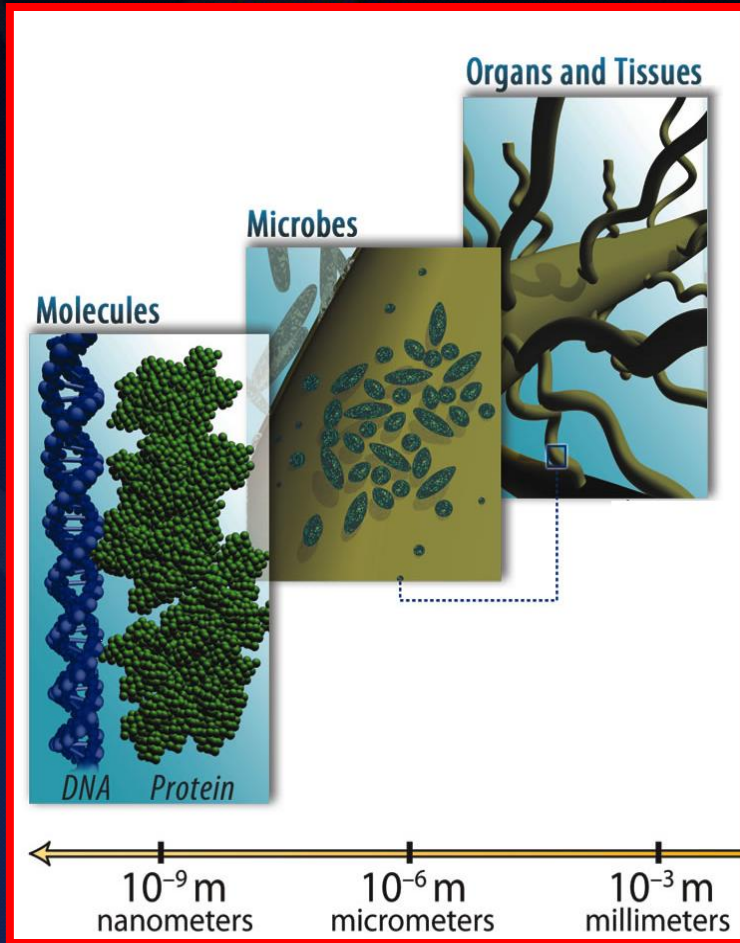


structure or function?
desired temporal resolution
desired spatial resolution
desired field of view
tissue culture, tissue slice, or in vivo ?

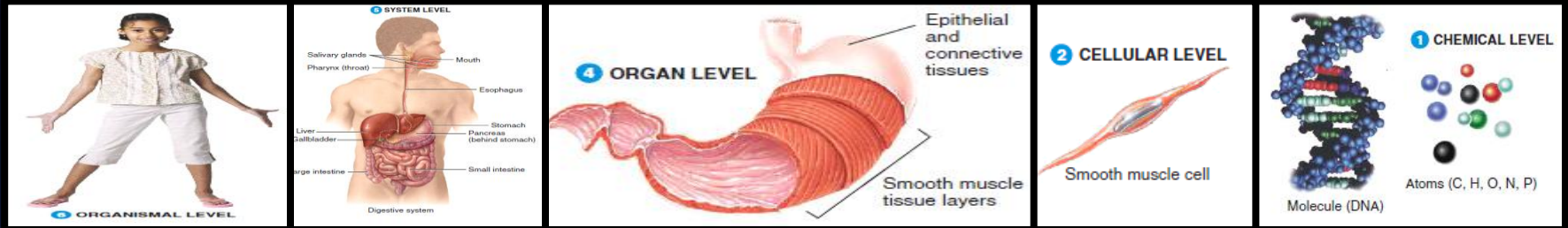
The Electromagnetic Spectrum



Spatial Scales



Spatial Scales in the Central Nervous System



CNS	Systems	Maps	Networks	Neurons	Synapses	Molecules
1 m	10 cm	1 cm	1 mm	100 μ m	1 μ m	0.1 nm

Positron Emission Tomography

Electron Microscopy

Ultrasound Imaging

Light Microscopy

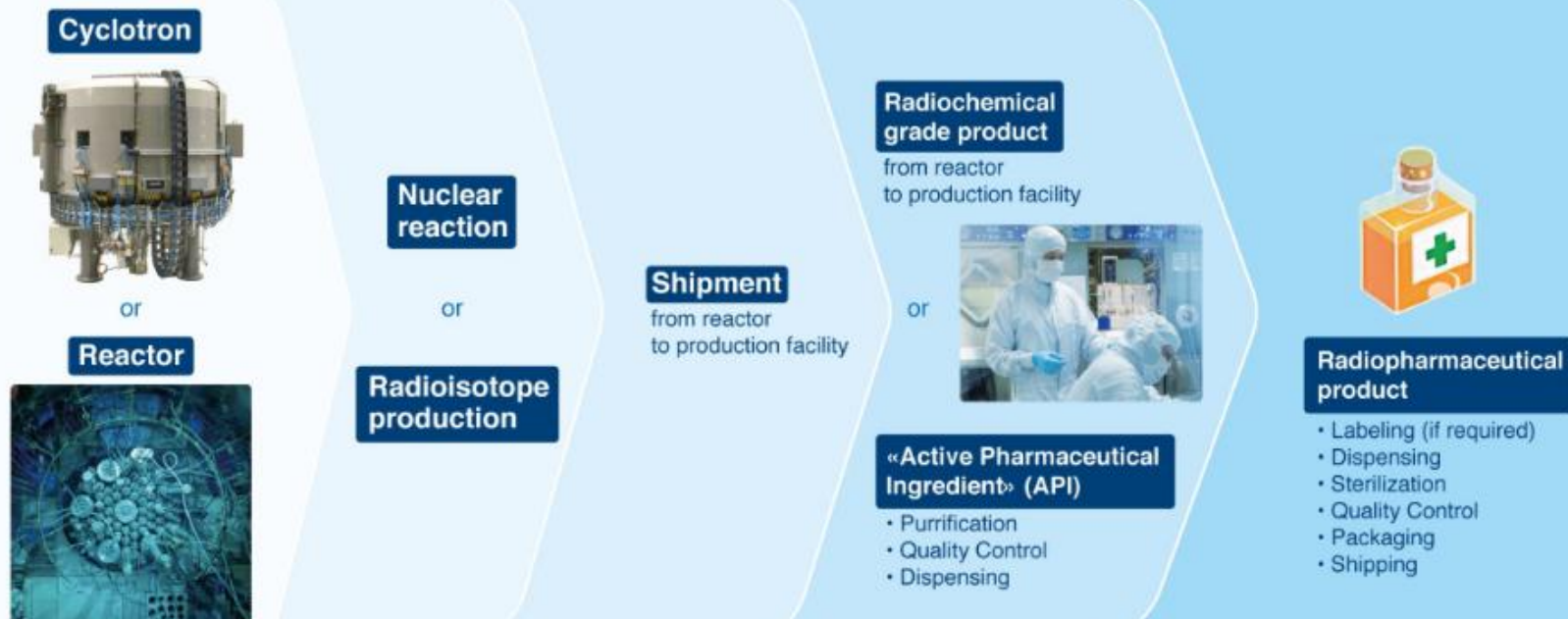
Autoradiography

Computed Tomography

Magnetic Resonance Imaging

Artificial Produced Radionuclides & Radiopharmaceuticals

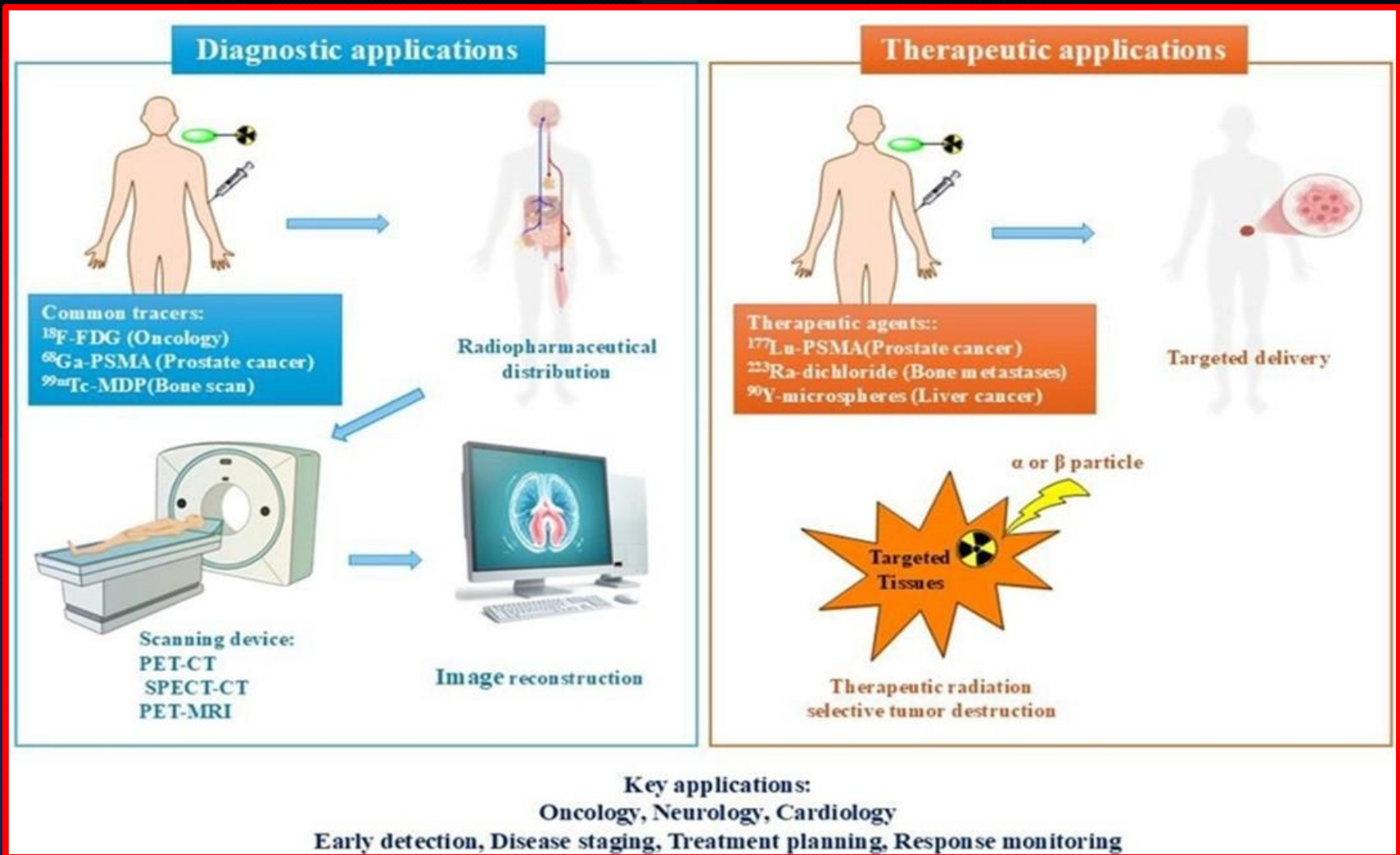
Radiopharmaceuticals Production

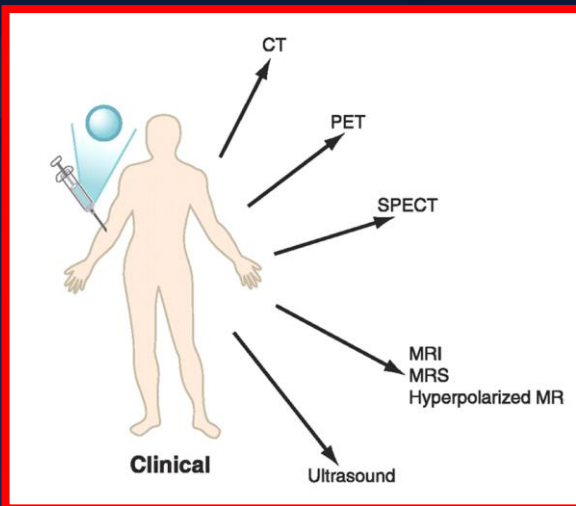
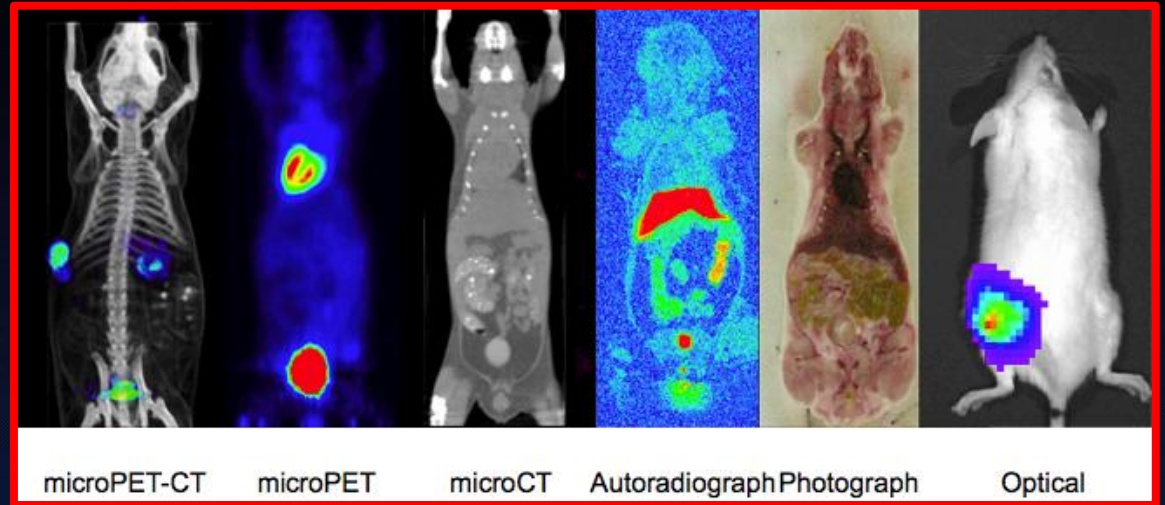
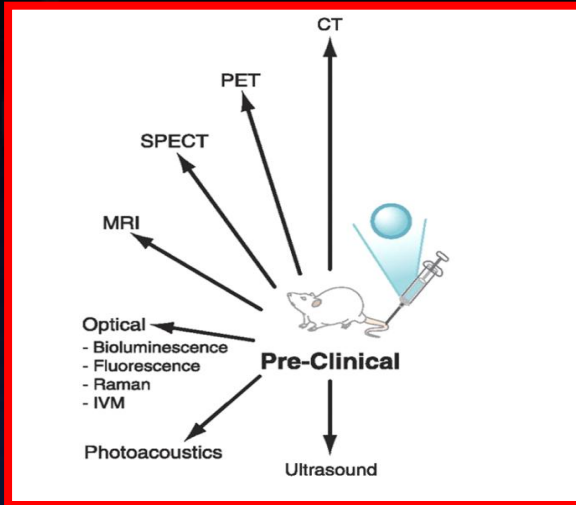


Radiopharmaceuticals Production



Radionuclides for Diagnosis and Therapy

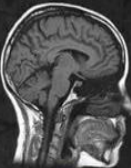




BRAIN IMAGING

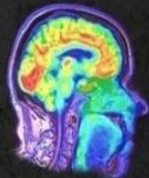


MRA
Magnetic Resonance Angiography



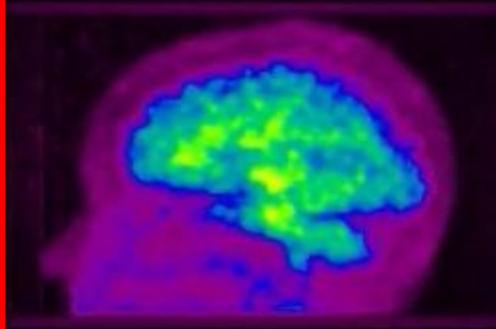
MRI
Magnetic Resonance Imaging

PET
SCAN
Positron Emission Tomography

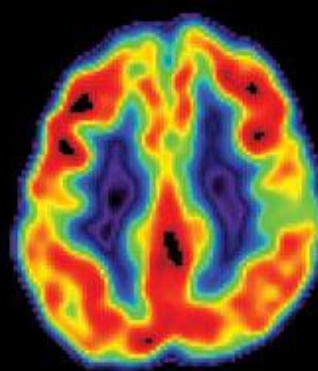
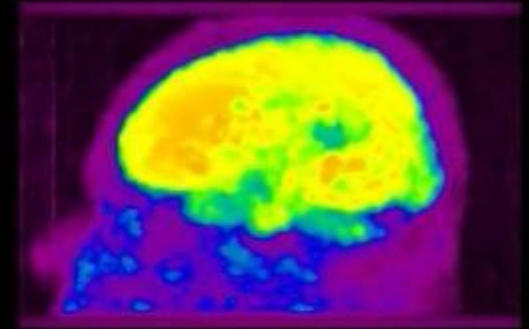


CT
Computed Tomography

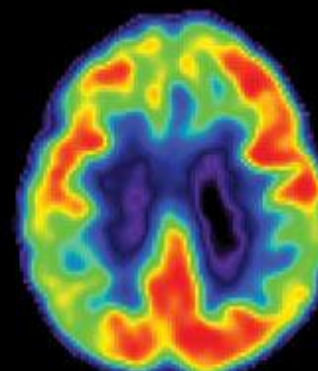
Normal brain



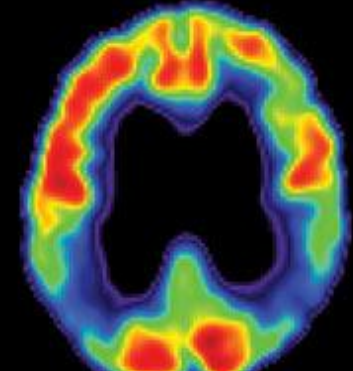
Alzheimer's brain



Normal

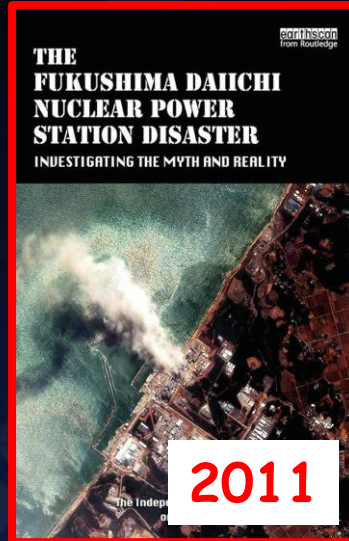
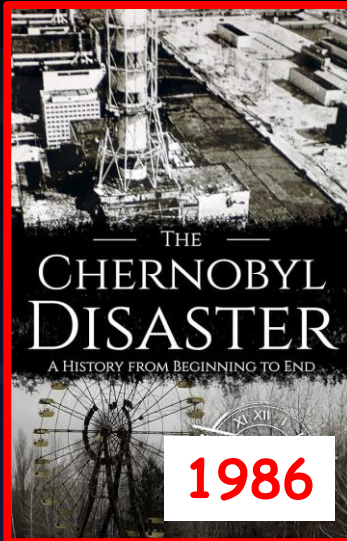
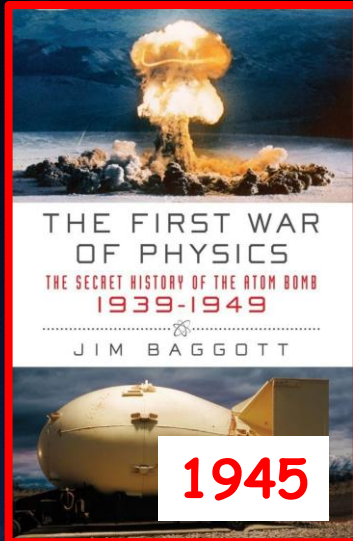


Mild cognitive
impairment



Alzheimer's
disease

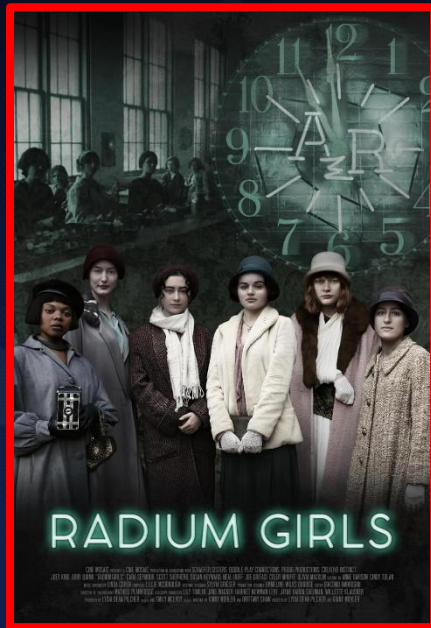
20th Century Age of (Ionising) Radiation



Examples of Collective Exposure

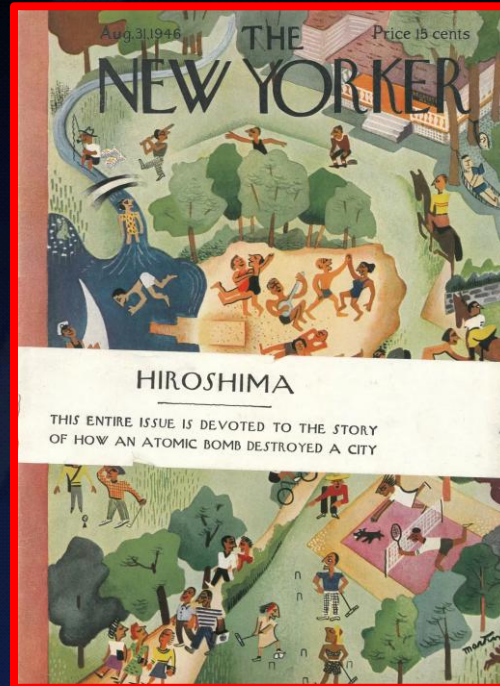
□ 1879

Uranium mijnwerkers
Erzgebirgste (CZ/DE)



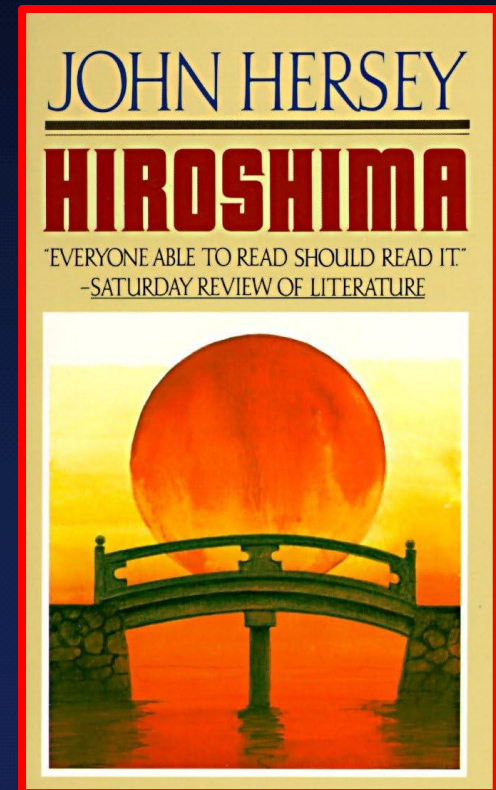
□ 1920s

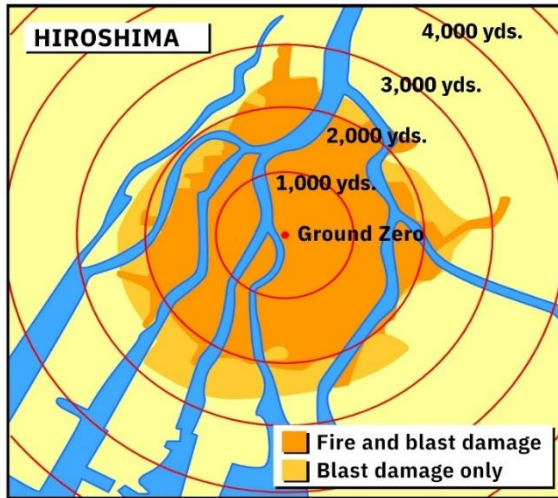
"Radium Girls" de wijzerplaat verfers
(lick, dip, paint)



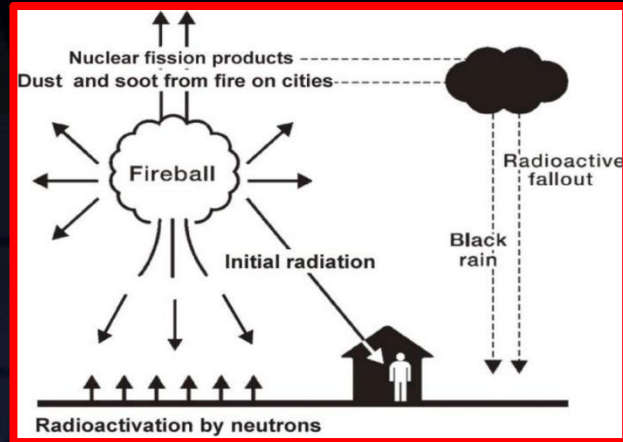
□ 1945

Hiroshima





Atomic Bomb Damage of Hiroshima

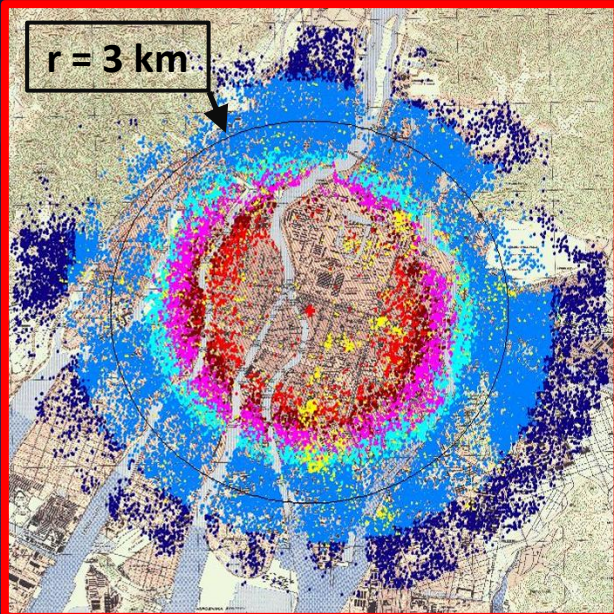


□ korte termijn effecten (1):

- explosie 50% energie
- hitte 35% energie
- straling 15% energie
- houten structuren verbrand < 2.0 km
- massale destructieve schokgolf < 2.5 km

□ korte termijn effecten (2):

- Hiroshima: 110.000 -140.000 doden
- 30%-40% van de populatie
- mortaliteit >60% binnen 1 km



Radiation exposure How does it compare?

Exposure measured in mSv

10,000

Fatal within weeks

LD100

6,000

Typical dosage recorded in those Chernobyl workers who died within a month

5,000

Single dose which would kill half of those exposed to it within a month

1,000

Single dose which could cause radiation sickness, nausea, but not death

400

Max radiation levels recorded at Fukushima plant 14 March, per hour

350

Exposure of Chernobyl residents who were relocated

100

Recommended limit for radiation workers every five years

10

Dose in full-body CT scan

9

Airline crew NYC -Tokyo polar route, annual

2

Natural radiation we're all exposed to, per year

1.02

Radiation per hour detected Fukushima site, 12 March

0.4

Mammogram breast x-ray

0.1

Chest x-ray

0.01

Dental x-ray

SOURCE: WNA, RADIOLOGYINFO.ORG, REUTERS

Biologic Effects based on Hiroshima/Nagasaki

DETERMINISTISCHE EFFECTEN

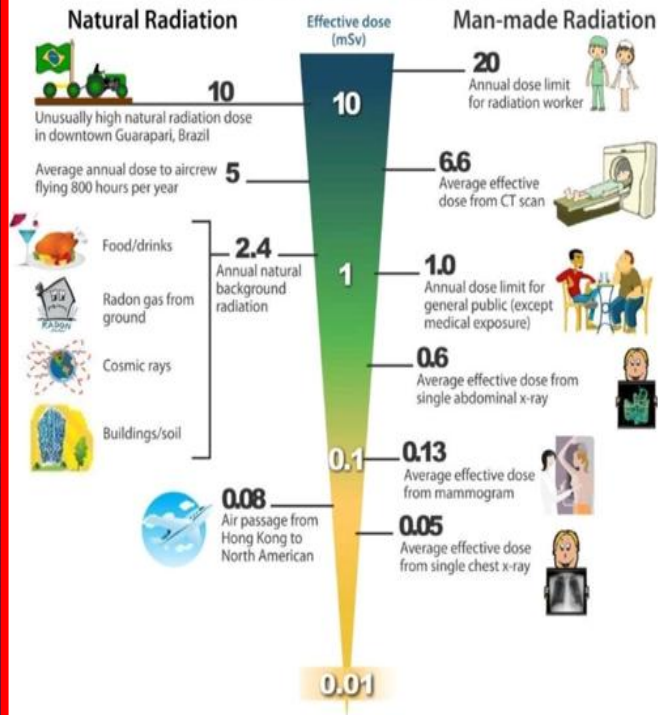
- Stralingsziekte
 - “Beenmerg” Syndroom
 - “Darm” Syndroom
 - “Hersen” Syndroom

STOCHASTISCHE EFFECTEN

- Risico op kanker

Radiation in Daily Life I

Radiation in Daily Life



Public exposure to natural radiation^a

Source of exposure		Annual effective dose (mSv)	
		Average	Typical range
Cosmic radiation	Directly ionizing and photon component	0.28	
	Neutron component	0.10	
	Cosmogenic radionuclides	0.01	
Total cosmic and cosmogenic		16%	0.39
External terrestrial radiation	Outdoors	0.07	
	Indoors	0.41	
Total external terrestrial radiation		20%	0.48
Inhalation	Uranium and thorium series	0.006	
	Radon (Rn-222)	1.15	
	Thoron (Rn-220)	0.1	
Total inhalation exposure		53%	1.26
Ingestion	K-40	0.17	
	Uranium and thorium series	0.12	
Total ingestion exposure		16%	0.29
Total		Public exposure to natural radiation 100%	2.4 mSv
			1.0-13

Radiation in Daily Life II

5 Radioactive Products ...We Use Every Day

Radioisotopes are unstable atoms that decay and emit excess energy, also known as radiation. There are hundreds of uses for radioisotopes and they can be found in countless everyday products—including these five for starters.

1

Clocks and Timepieces

Some wrist watches contain a gaseous tritium light source (GTLS), which are sealed glass containers filled with radioactive tritium gas that is used to make a timepiece glow in the dark.

2

Gemstones

Irradiated gemstones include amethysts, yellow sapphires, and green diamonds, and their initial distribution is actually regulated by the U.S. Nuclear Regulatory Commission.

3

Smoke Detectors

Household detectors contain a tiny amount of Americium-241, a radioactive isotope used to detect smoke particles. The two charged plates inside create a flow of positively and negatively charged ions that trigger the alarm when smoke breaks the constant flow of ions.

4

Exit Signs

Some exit signs contain tiny traces of tritium, a rare and radioactive isotope of hydrogen, which is why they're able to glow in the dark without batteries. The tritium is contained in sealed glass tubes lined with phosphor. Low-energy beta particles emitted by the tritium bombard the phosphor, causing it to glow.

5

Snack Time

Bananas contain naturally occurring radionuclides — radioactive potassium-40, to be exact — which, according to the EPA, means they can emit .01 millirem (0.1 microsieverts). And if you attend that happy hour, there's a teeny bit (about 390 Picocurie per kilogram) of the same isotope potassium-40 in your beer too!



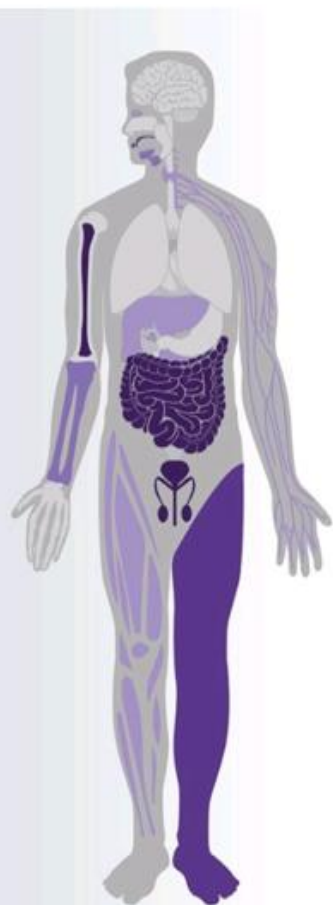
LEARN MORE AT energy.gov/ne

U.S. DEPARTMENT OF
ENERGY

Office of
NUCLEAR ENERGY

Radiation Dose from Medical Examinations

Organs & tissues sensitivity to ionizing radiation



HIGH

Bone marrow
Reproductive cells
Intestines
Lymphoid tissue

MODERATELY HIGH

Oral mucosa
Skin

MODERATELY LOW

Salivary glands
Mature bone
Mature cartilage
Thyroid gland tissue

LOW

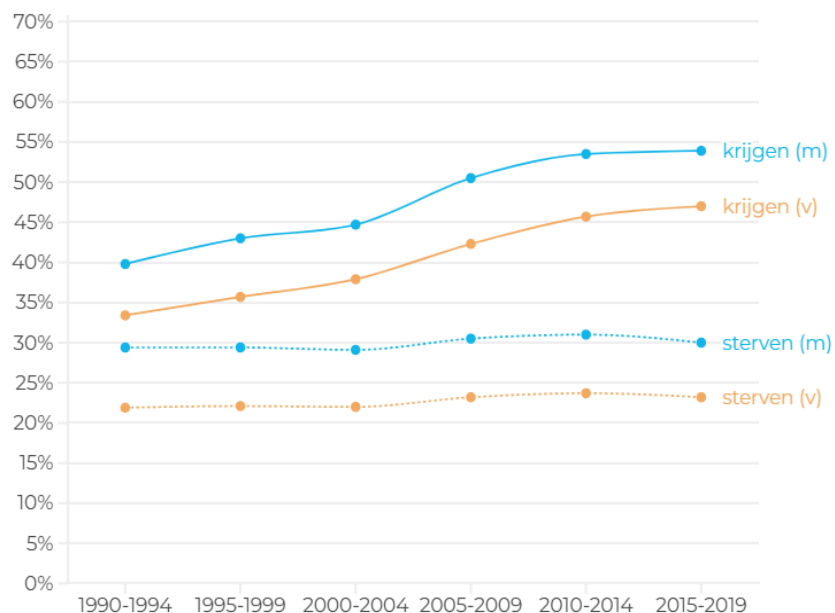
Liver
Optic lens
Kidneys
Muscle
Nerve

Radiation Dose to Adults From Common Imaging Examinations

	Procedure	Approximate effective radiation dose	Comparable to natural background radiation for
 ABDOMINAL REGION	Computed Tomography (CT) — Abdomen and Pelvis	10 mSv	3 years
	Computed Tomography (CT) — Abdomen and Pelvis, repeated with and without contrast material	20 mSv	7 years
	Computed Tomography (CT) — Colonography	6 mSv	2 years
	Intravenous Pyelogram (IVP)	3 mSv	1 year
	Barium Enema (Lower GI X-ray)	8 mSv	3 years
	Upper GI Study With Barium	6 mSv	2 years
 BONE	Spine X-ray	1.5 mSv	6 months
	Extremity (hand, foot, etc.) X-ray	0.001 mSv	3 hours
 CENTRAL NERVOUS SYSTEM	Computed Tomography (CT) — Head	2 mSv	8 months
	Computed Tomography (CT) — Head, repeated with and without contrast material	4 mSv	16 months
	Computed Tomography (CT) — Spine	6 mSv	2 years
 CHEST	Computed Tomography (CT) — Chest	7 mSv	2 years
	Computed Tomography (CT) — Lung Cancer Screening	1.5 mSv	6 months
	Chest X-ray	0.1 mSv	10 days
 DENTAL	Dental X-ray	0.005 mSv	1 day
 HEART	Coronary Computed Tomography Angiography (CTA)	12 mSv	4 years
	Cardiac CT for Calcium Scoring	3 mSv	1 year
 MEN'S IMAGING	Bone Densitometry (DEXA)	0.001 mSv	3 hours
 NUCLEAR MEDICINE	Positron Emission Tomography — Computed Tomography (PET/CT)	25 mSv	8 years
 WOMEN'S IMAGING	Bone Densitometry (DEXA)	0.001 mSv	3 hours
	Mammography	0.4 mSv	7 weeks

Risk of Cancer/Dying in a Lifetime

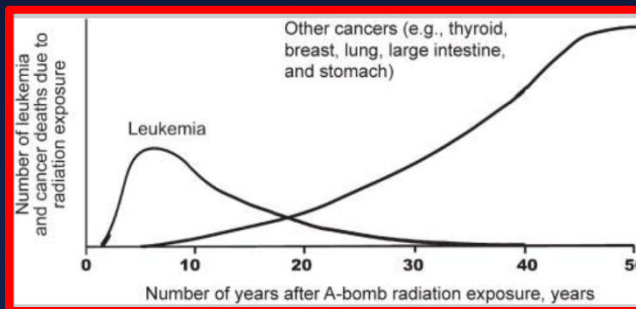
De kans op het krijgen van/sterven aan kanker



KNL integraal
kankercentrum
Nederland

Het risico op kanker ten
gevolge van
ioniserende straling is:

~ 5 % per Sievert



<https://www.ntvg.nl/artikelen/ontwikkeling-van-de-kans-op-kanker-1990-2019>

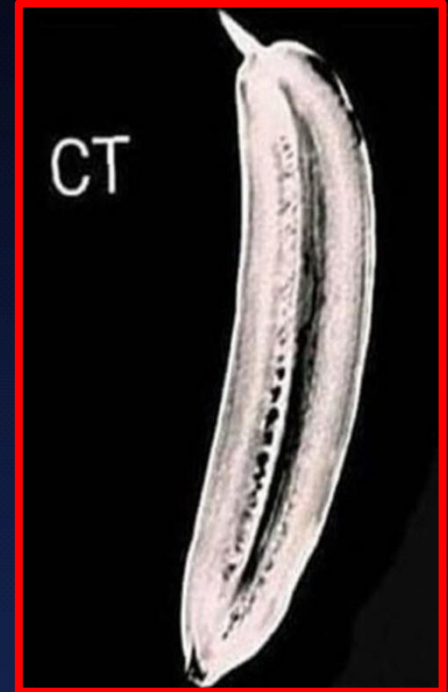
Non-Medical Imaging



- Drug- and Human Trafficking
- Baggage Screening
- Art Objects

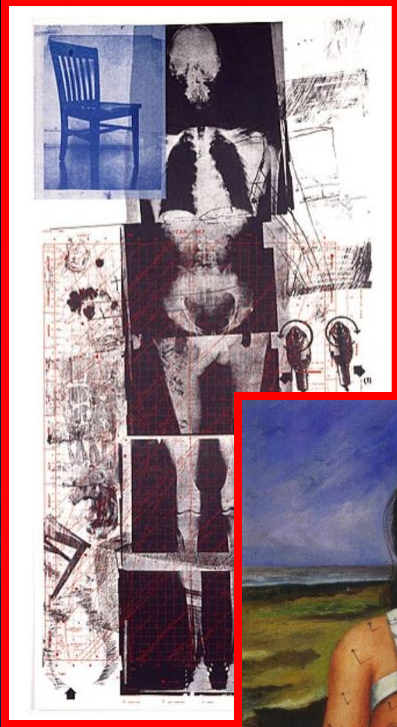


Non-Medical Imaging

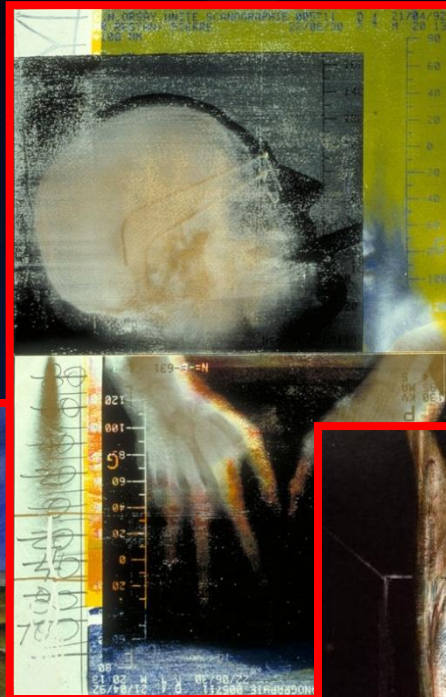


Non-Medical Imaging and the Arts

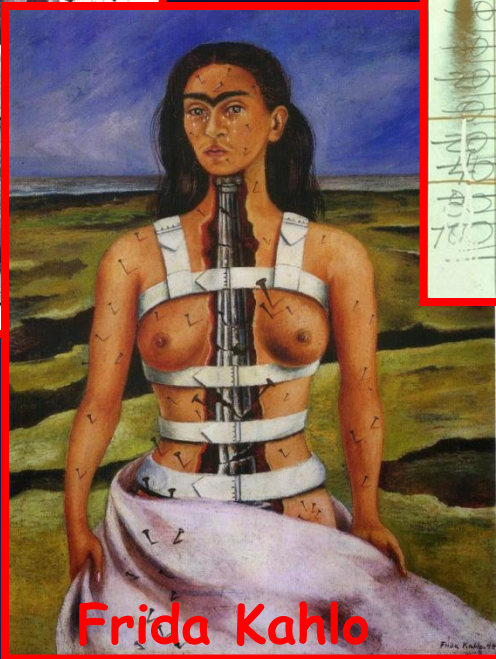
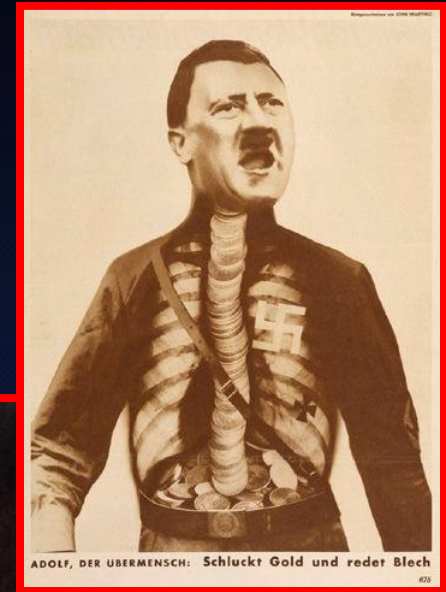
Rauschenberg



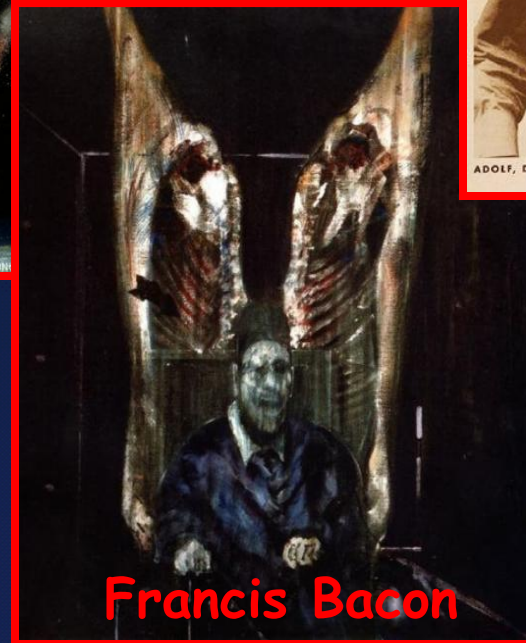
Steve Miller



Heartfield

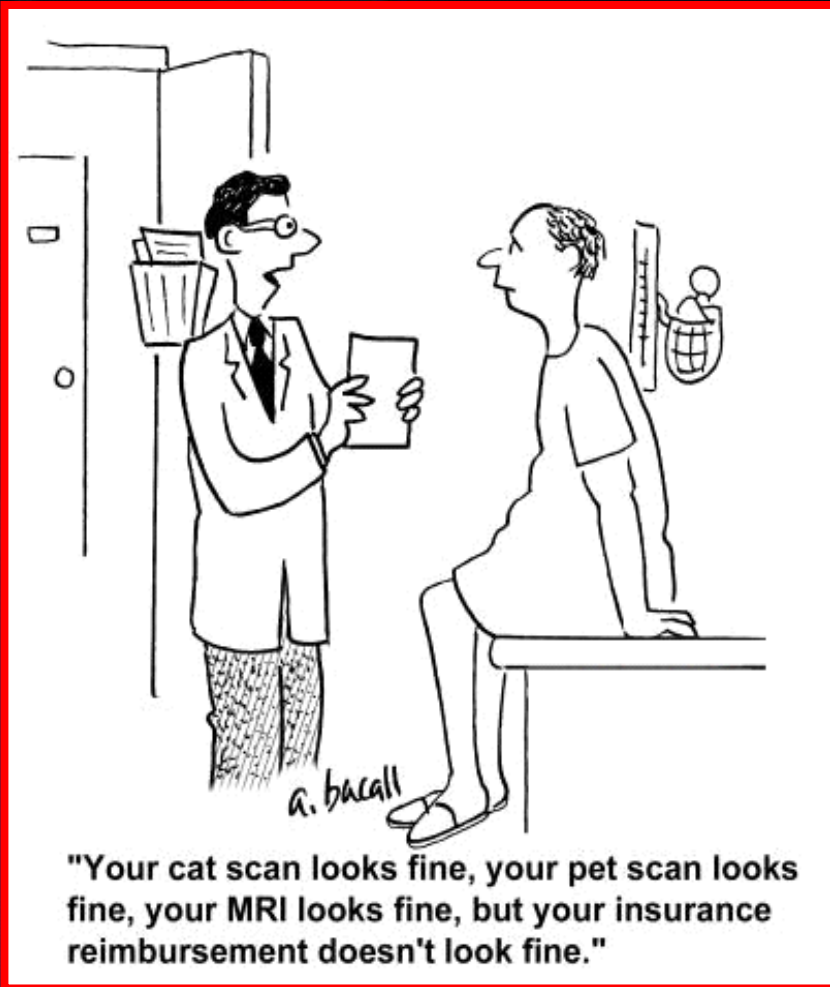


Frida Kahlo



Francis Bacon

Some Medical Related Topics



- Preventive Whole-Body Scan?
- Unjustified use of Medical Procedures
- Overall Increase of Collective Dose
- Personalised Medicine
- Health Insurance costs

Some Medical Related Conclusions



- Medical Imaging is a non-invasive (mostly diagnostic) practice.
- Medical Imaging is team work and needs highly qualified personnel. (High Tech and Multi-Discipline)
- Medical Imaging Modalities are complementary.
- Medical Imaging has a spin off in many domains.
- Overall Increase of Collective Dose
- Danger of Medicalization

Some General Conclusions

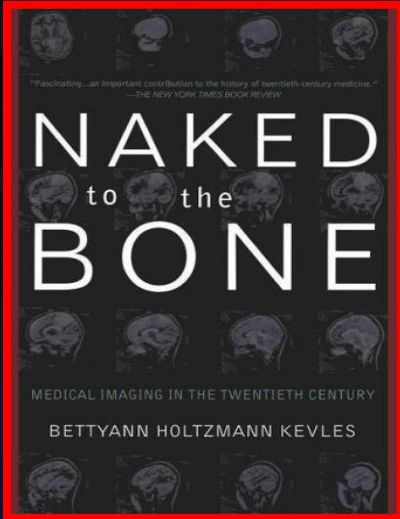


Francisco de Holanda (1545)

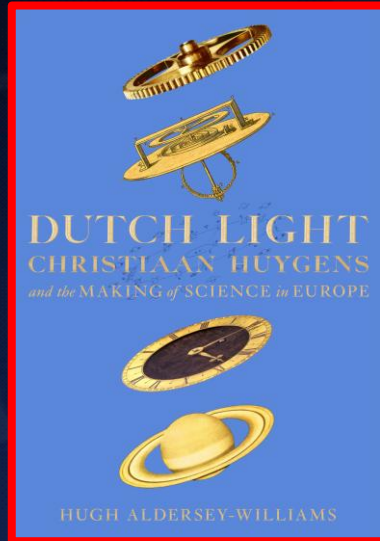
- Radiation is a General Natural Phenomenon all around in Time and Space.
- In Short:
Radiation is Energy on the move.
- In Practice:
Be aware of the distinction between Ionising and Non-Ionising Radiation

Exposure to Ionising Radiation is basically not without Risk.

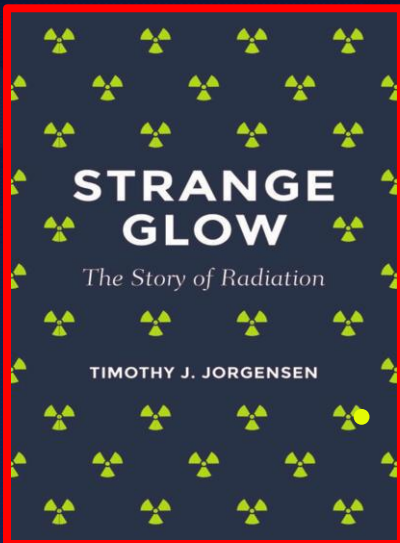
References



Naked to the Bone: Medical Imaging in the Twentieth Century (1997) - *B. Kevles*



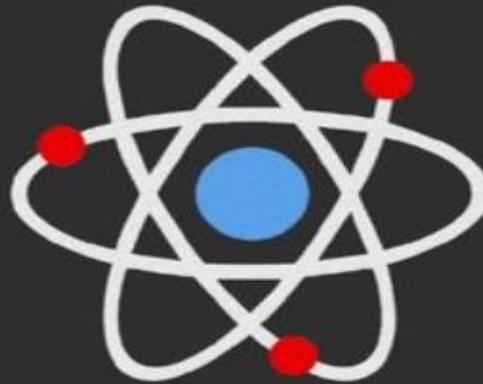
Dutch Light (2020) -
Hugh Aldersey-Williams



Strange Glow (2020) -
Timothy J. Jorgensen

- YouTube (DWDD Licht)
- <http://www.unscear.org/>
- <https://www.iaea.org>
- <http://www.who.int>

Think like a
Proton



and stay
Positive